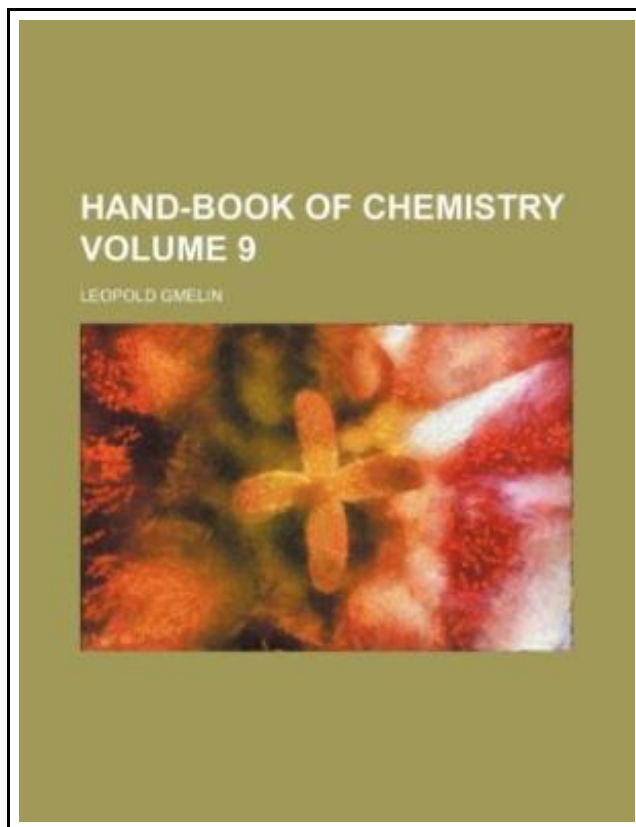


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1855 Excerpt: . lead. (Dessaignes.)--When glycolic acid is dissolved in hot alcohol, sulphuric acid added by drops to the solution when cold, and the mixture set aside for a few days, crystals (a) are obtained,  $C_4H_5O_3$ ,  $SO_3$ , sometimes in long slender prisms with right terminal faces, sometimes in tables having a strong lustre. The crystals have a sour taste; are permanent in the air; do not lose weight at  $100^\circ$ ; dissolve in water and in warm aqueous alcohol, but not in absolute alcohol or in ether.--Proceeding in the same manner, with the exception of heating to the boiling point after the addition of sulphuric acid, the crystals  $C_4H_5O_3$ ,  $SO_3$  are sometimes obtained in a form resembling that of common sulphate of copper (Horsford). 115. 100-00 The large quantity of hydrogen obtained in a renders the analysis doubtful. Horsford likewise obtained the two following basic compounds: Disulphate of Glycolic acid.--The crystals contain 25-70 per cent of C and 60.1 H, and therefore consist of  $2C_4H_5O_3$ ,  $SO_3$ . Sesquibasic Sulphate of Glycolic acid.--a. The crystals contain 27.74 p.c. sulphuric acid, and therefore consist of  $2C_4H_5O_3$ ,  $SO_3$ .--ft. A mixture of the solutions of - and y yields crystals, containing 25-65 p.c. sulphuric acid and therefore  $2C_4H_5O_3$ ,  $SO_3$ ,  $HO$ .--7. A solution of glycolic acid in weak spirit, mixed with sulphuric acid even in large excess, deposits in the course of 24 hours, long rectangular prisms which are acid and permanent in the air (Horsford).  $3C_4H_5O_3$ ,  $2SO_3$ ,  $2HO$ . 323. 100-00 100 00 Hydrochloric acid, or...



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